

# Air Monitoring Instrumentation Nitrogen Oxides (NO<sub>y</sub>)

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***David A. Neuschuler***

*Director of Engineering*

*Teledyne – API*

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# What is NO<sub>y</sub>

- Total Reactive Nitrogen

"Collective name for oxidized forms of nitrogen in the atmosphere such as nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), nitric acid (HNO<sub>3</sub>), and organic nitrates; usually designated by NO<sub>y</sub>" - *AMS*

- Precursors in the formation of Ozone

- Definitions

$$\text{NO}_y = \text{NO}_z + \text{NO}_x$$

$$\text{NO}_x = \text{NO} + \text{NO}_2$$

$$\begin{aligned} \text{NO}_z = & \text{HNO}_3 + \text{HONO} + 2\text{N}_2\text{O}_5 + \text{HO}_2\text{NO}_2 \\ & + \text{PAN} + \text{NO}_3 + \text{Organic Nitrates} - \text{but } \mathbf{not} \text{ NH}_3 \end{aligned}$$

- Some NO<sub>z</sub> compounds have short lifetimes

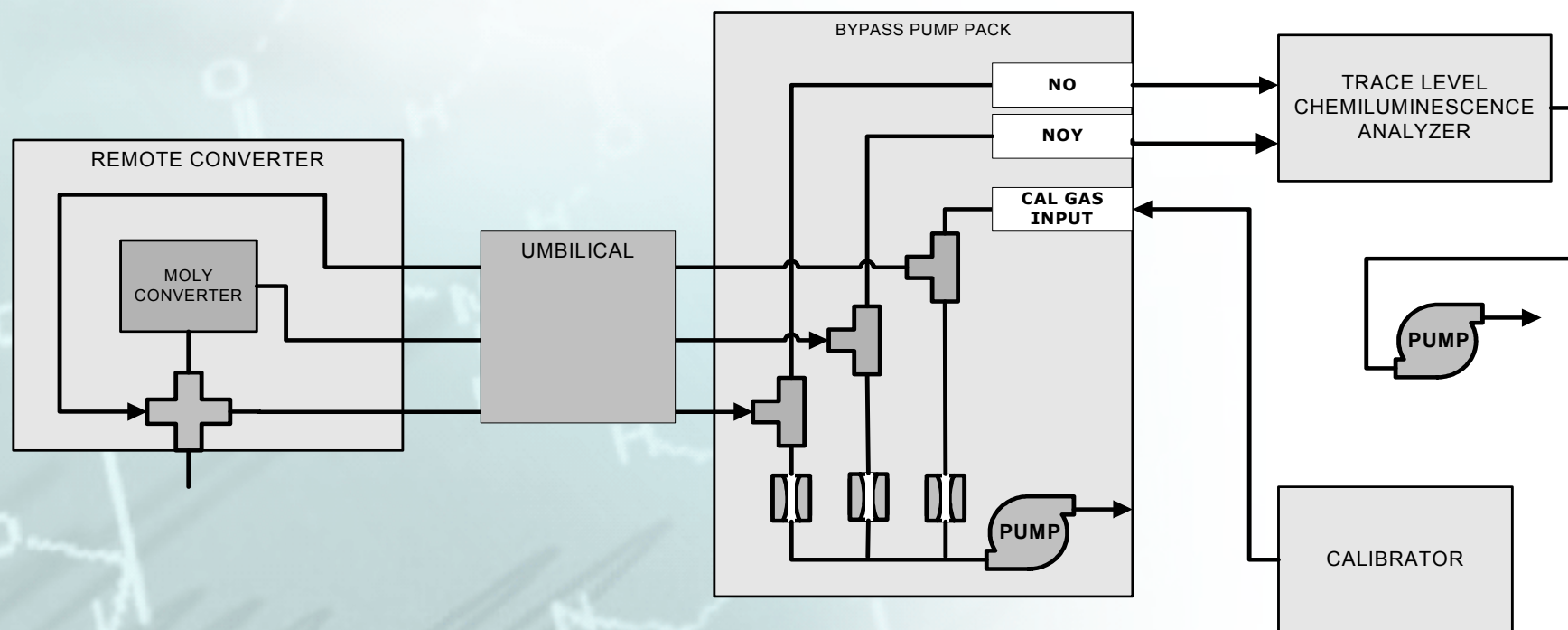
- NO<sub>2</sub> specific analyzer required to measure NO<sub>z</sub>

# Technique

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- Measurement using Nitric Oxide-Ozone Chemiluminescence analyzer
- Conversion of NO<sub>y</sub> species to NO
  - Molybdenum ~325°C
  - Gold with CO or H<sub>2</sub> injection ~ 400°C
  - Vitreous Carbon ~ 350°C
  - Ferrous Sulfate
- Converter as near inlet as possible with no sample filter
- Requires trace level analyzer for useful measurement
- Minimizing residence time essential to good measurement

# Simple Pneumatic Block Diagram





# Analyzer Differences

Specification or Characteristic	Standard NOx Analyzer	Premium NOx Analyzer	Purpose
LDL	<400 ppt	<50 ppt	
zero drift 24 hours	<500 ppt	<100 ppt	
zero drift 7 days	<1000 ppt	<200 ppt	
Propylene reject. Ratio		>20,000:1	
Ethylene reject. Ratio		>40,000:1	
PMT Anode Sensitivity	2500 A/lm	3000 A/lm	More sensitivity, higher SNR
Reaction Cell	Non-plated	Gold Plated	Increased signal out
Flow Rate	500 ccm	1000 ccm	Increased signal out
Nominal Cell Pressure	5" HgA	2.5" HgA	Increased sensitivity
Autoref scheme	Autozero stop sample flow	Autozero with prereactor	Improved hydrocarbon rejection

# Inlet Materials

(or how can we keep **"It"** from sticking)

- **"It"** = gasses that readily stick to surfaces especially nitric acid and ammonia
- **Causes** memory effect leading to excessive rise and fall times:
  - Operation – over or under actual value, smear features
  - Calibration – Elevated zeros, reduced spans
- Exacerbated by long tubing, low temperatures

# Inlet Materials

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Standard: ¼" stainless tubing, adapters and fittings

Field Testing:

- SilcoSteel® (hydrogenated amorphous silicon) coated, stainless for inlet and other NO<sub>y</sub> wetted surfaces, including body and inlet tubing of converter
- PFA bulkhead, cross and tubing to within ½" of converter inlet; converter inlet and body SilcoSteel® coated

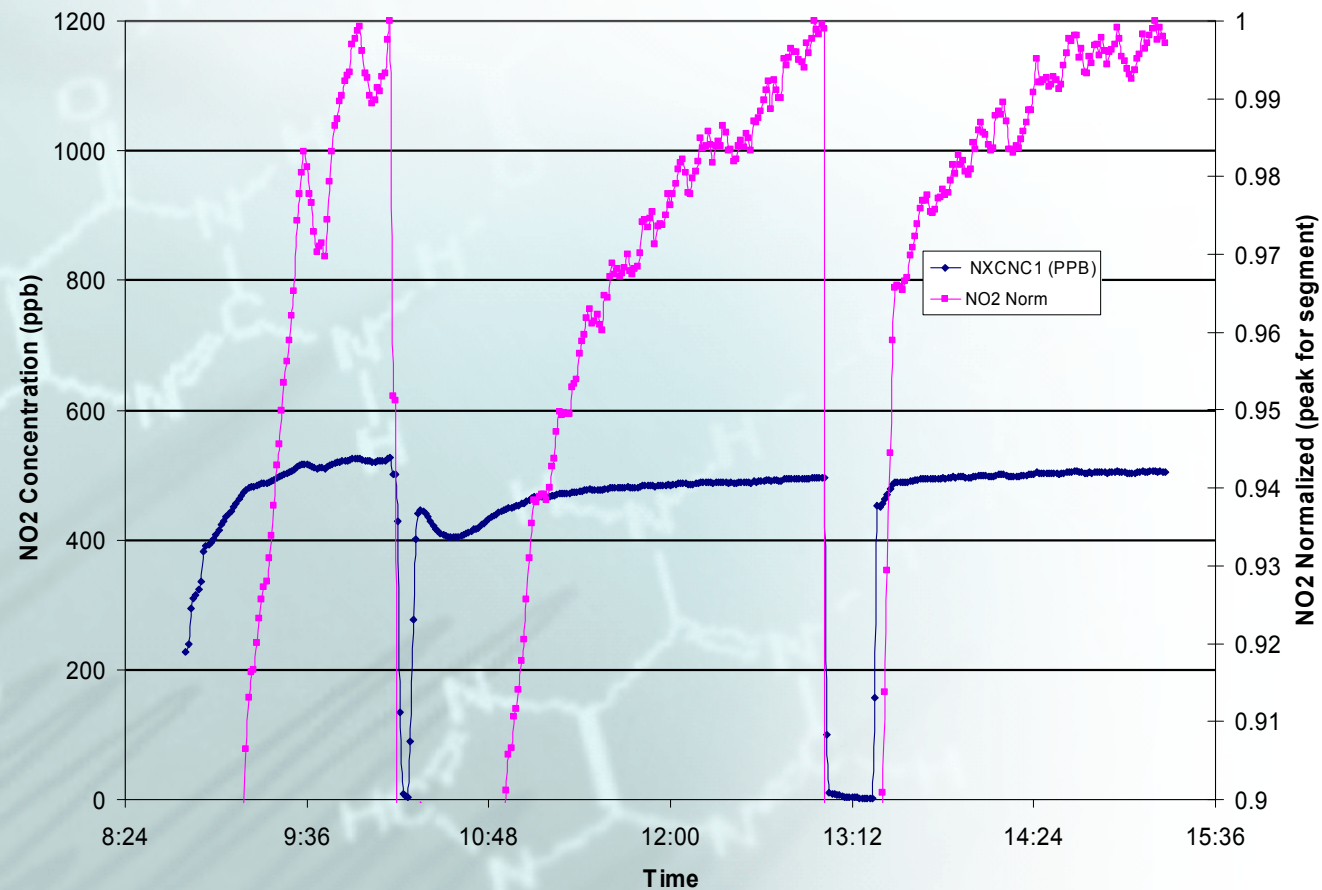
# Calibration

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- Gases: Nitric, iPan, nPan, NO<sub>2</sub>, NO/GPT
  - No SRM
  - Nitric: very difficult to use
  - N-propyl nitrate: difficult to obtain & questionable analog for nitric
  - Iso-propyl nitrate: readily available
- Errors in verifying conversion efficiency using bottles
- Big cal gas flows required
- Can not run from common manifold
- Losses due to conditioning
- Must be VERY, VERY patient



## Conditioning with 500 ppb NO<sub>2</sub>



# Checking Converter Efficiency

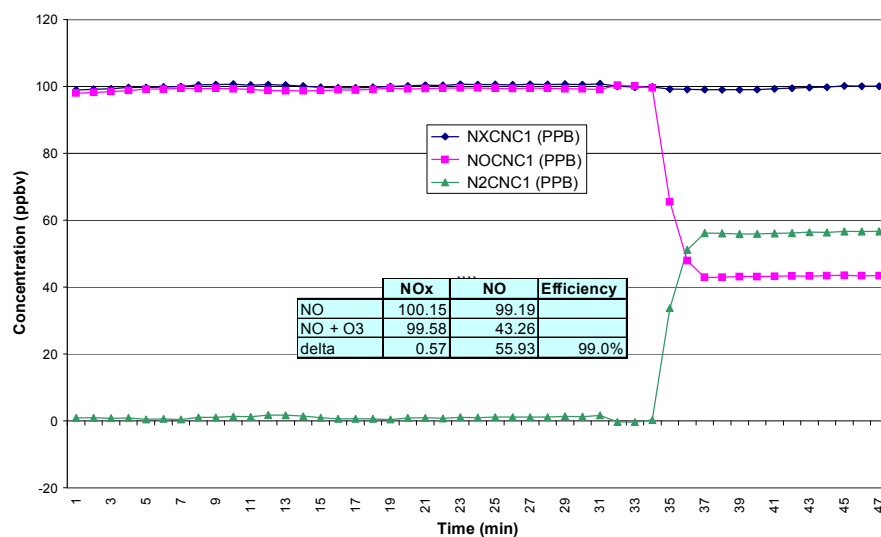
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- Gas Phase Titration (GPT) golden standard
  - Auto-referencing - independent of NO or Ozone concentration
  - Independent of MFC calibrations and matching
- NO/NO<sub>2</sub>, nPan or i-Pan bottles
  - Limited by accuracy of bottles
  - Limited by accuracy and linearity of MFCs
  - Best to maintain MFC flows use ratio of bottles
  - Check for contaminants in NO and other bottle

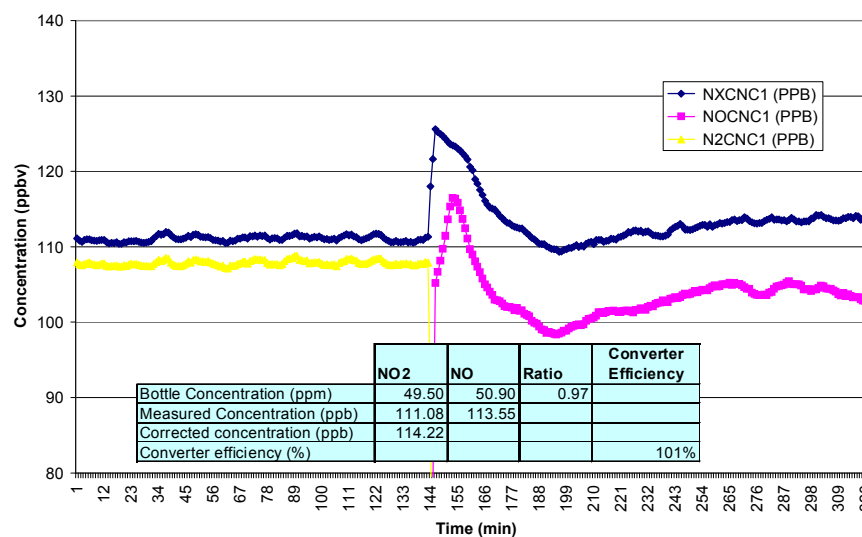
# GPT vs. NO/NO2 gas Efficiency

Method	Efficiency
GPT	99%
Gas Bottles	101%

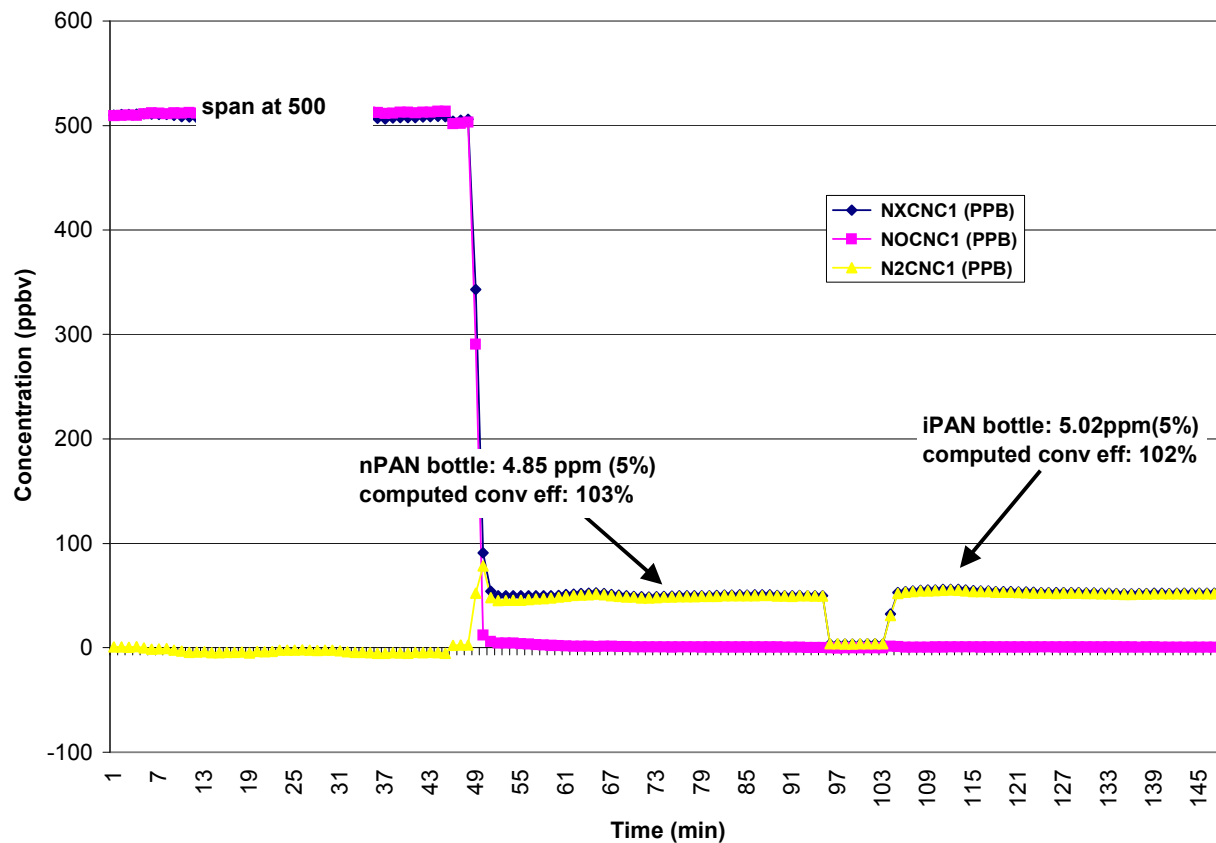
Efficiency by GPT (100 ppbv)



Efficiency with NO & NO2 Bottles



# iPan/nPan Efficiency





# Interferences

- Water: no effect at zero, ~ 3% quench at span
- $\text{NH}_3$ : 1 – 5 %
- Ammines ??
- Hydrocarbons

\* WATER INTERFERENCE DATA - TABLE 14

DATA	TEST 1	TEST 2	TEST 3	TEST 4	TEST 5	TEST 6	TEST 7
R4	104.9 ppb	104.3 ppb	104.1 ppb	106.2 ppb	106.4 ppb	106.1 ppb	99.4 ppb
R14	99.1	98.7	99.0	100.7	101.2	100.6	95.1
R14'	101.3	100.9	101.2	102.9	103.4	102.8	97.2
IE	-3.6	-3.4	-2.9	-3.3	-2.7	-3.3	-2.2

# Siting

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- Be aware of nearby sources of:
  - Hydrocarbons – roofing materials
  - Ammonia – sewer vents



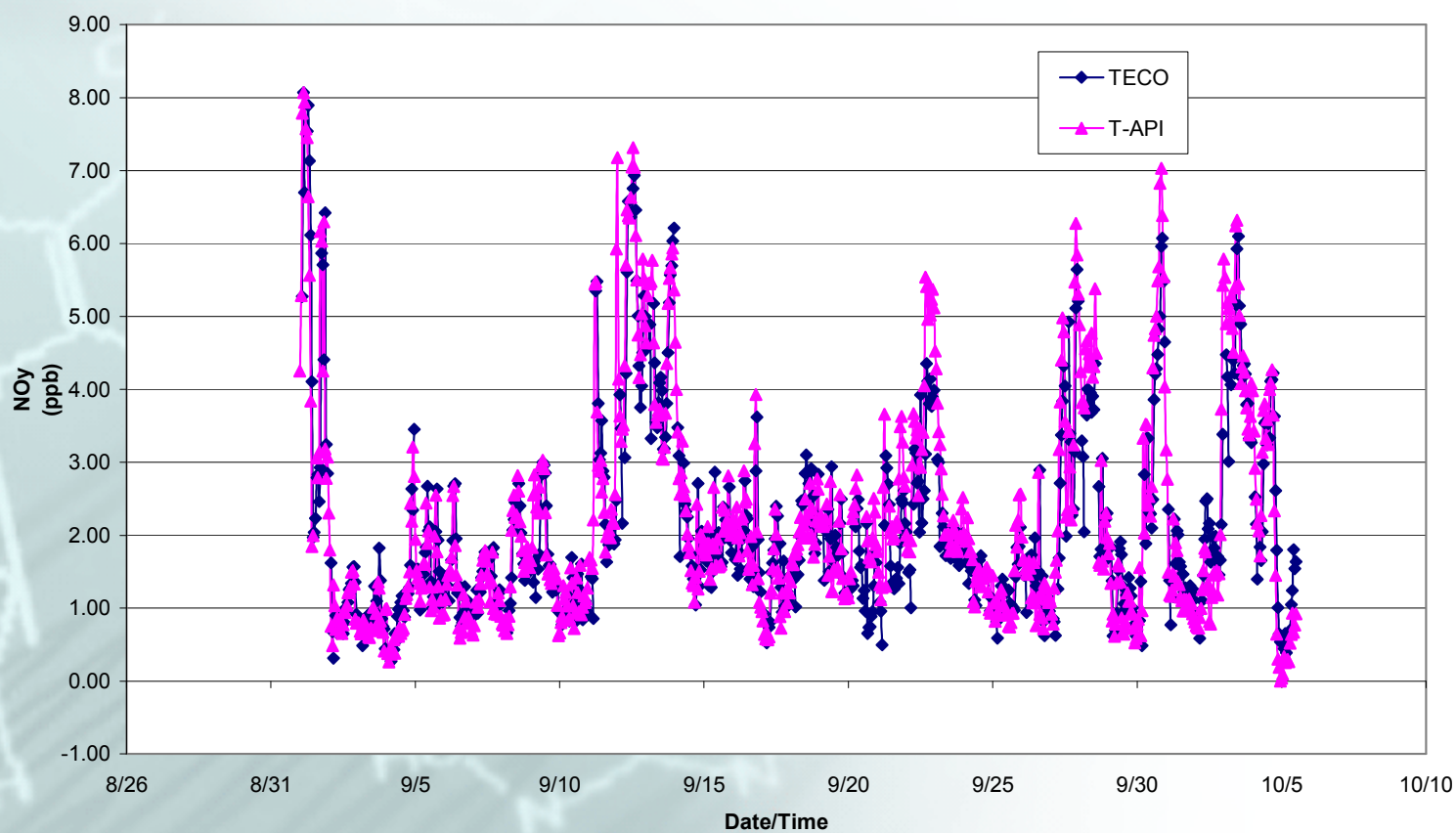
NSF UV Spectroradiometer – Barrow, AK  
Courtesy of Biospherical Instruments



Pinnacle State Park, NY

# Pinnacle State Park NO<sub>y</sub> Comparison

Courtesy of Atmospheric Sciences Research Center  
University at Albany - State University of NY



Air Monitoring Instrumentation - Nitrogen Oxides (NO<sub>y</sub>)

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# Acknowledgements & Bibliography

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- **Review of M200AU: NO<sub>y</sub> Converter Design Theory and Practice**  
Martin Buhr, Regional Air Quality Council, Denver, CO, 1997
- **AMS *Glossary of Meteorology***  
American Meteorological Society  
<http://amsglossary.allenpress.com/glossary/acknowledge>